Scientific term	Definition	Example(s) or Diagram
materials	A material is any substance that has a name, eg chalk, paper, wood. Everything is made up of materials and we can choose the best material for the job.	
solid	The shape of a solid does not change on its own. It is rigid and has a fixed volume.	
liquid	The shape of a liquid does change, it is not rigid. It fits the shape of the container it is put in.	
gas	Gases do not have a shape. They completely fill any container they are put into. have the same volume as the container.	
molecules	A molecule is two or more atoms joined (or "bonded") tightly together.	
changing state	We can change a material's state by changing its temperature, eg. ice (solid) to water (liquid) to steam (gas) and in reverse.	
condensation	Water vapour (water in its gas form) cools and turns into liquid.	
evaporation	When a liquid is heated, it evaporates and becomes a gas.	
melting	Melting is the process of changing a solid into a liquid (like ice into liquid water).	
filtering	Filtering is a process by which impurities or particles are removed from either a liquid or a gas.	
dissolving	When a substance dissolves , it might look like it has disappeared, but in fact it has just mixed with the water to make a transparent (see-through) liquid called a solution.	
soluble	A solution is made when one substance "dissolves" into another and forms a solution. Larger groups of molecules break down into much smaller groups or individual molecules.	
insoluble	A solid's particles will not dissolve and combine with a liquid's particles.	

reversible change	This is when materials can be changed back to how they were before the reaction took place, eg. Ice melts to form water and can be frozen back to ice again.	
irreversible change	Irreversible Change is when materials cannot be changed back to how they were before.	
solidification	Solidification , also known as freezing, is when a solid is formed.	
sieving	Sieving is aprocess to separate finer particles from coarser ones or solids from liquids.	
chemical changes	A chemical change is any change that causes a new substance to be formed.	
physical changes	A physical change is a change in which no new substances are formed.	
reaction	A chemical reaction is a process where a set of substances undergo a chemical change to form a different substance.	
permeable	Means to allow liquids in particular to pass through a solid.	
transparent	Transparency is the property of allowing light to pass through something. An object that is transparent can be seen through.	
translucent	Translucent materials allow light to pass through but so the object is not clearly visible.	
opaque	Opaque materials do not allow any light to pass through. The object on the opposite side of the light is not visible at all.	
viscosity	Viscosity is the property of a liquid that describes how fast or slowly it will flow. You can think of viscosity as how thick a liquid is	
conduct	Conductors are substances that an electric charge can pass through without difficulty.	
insulate	Insulators are materials which do not conduct heat very well and so we can use them to control heat and keep things hot or cold.	
magnetism	The force of attraction or repulsion between substances made of certain materials, eg. iron, nickel, cobalt and steel.	

Word Bank: Changing Materials